

TOTAL SYNTHESIS OF ERGOT ALKALOID, (\pm)-6,7-SECOAGROCLAVINE¹

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Ergot alkaloid, (\pm)-6,7-secoagroclavine was synthesized from 2-methyl-5-nitroisoquinolinium iodide by three routes. In the course of the study, a novel intra-molecular π -alkylation of allyl alcohol was found. Reaction of Grignard reagents with nitroalkanes to afford N-alkyl hydroxylamines was effectively used in the present synthesis.

In the previous papers,² we described a two step synthesis of 4-(indol-4-yl)-3-buten-2-one (2) from 2-methyl-5-nitroisoquinolinium iodide (1) and its conversion to 5-acetonyl-1,3,4,5-tetrahydro-4-nitrobenz[cd]indole (4t and 4c in ca. 3:1 ratio) via Mannich base (3).

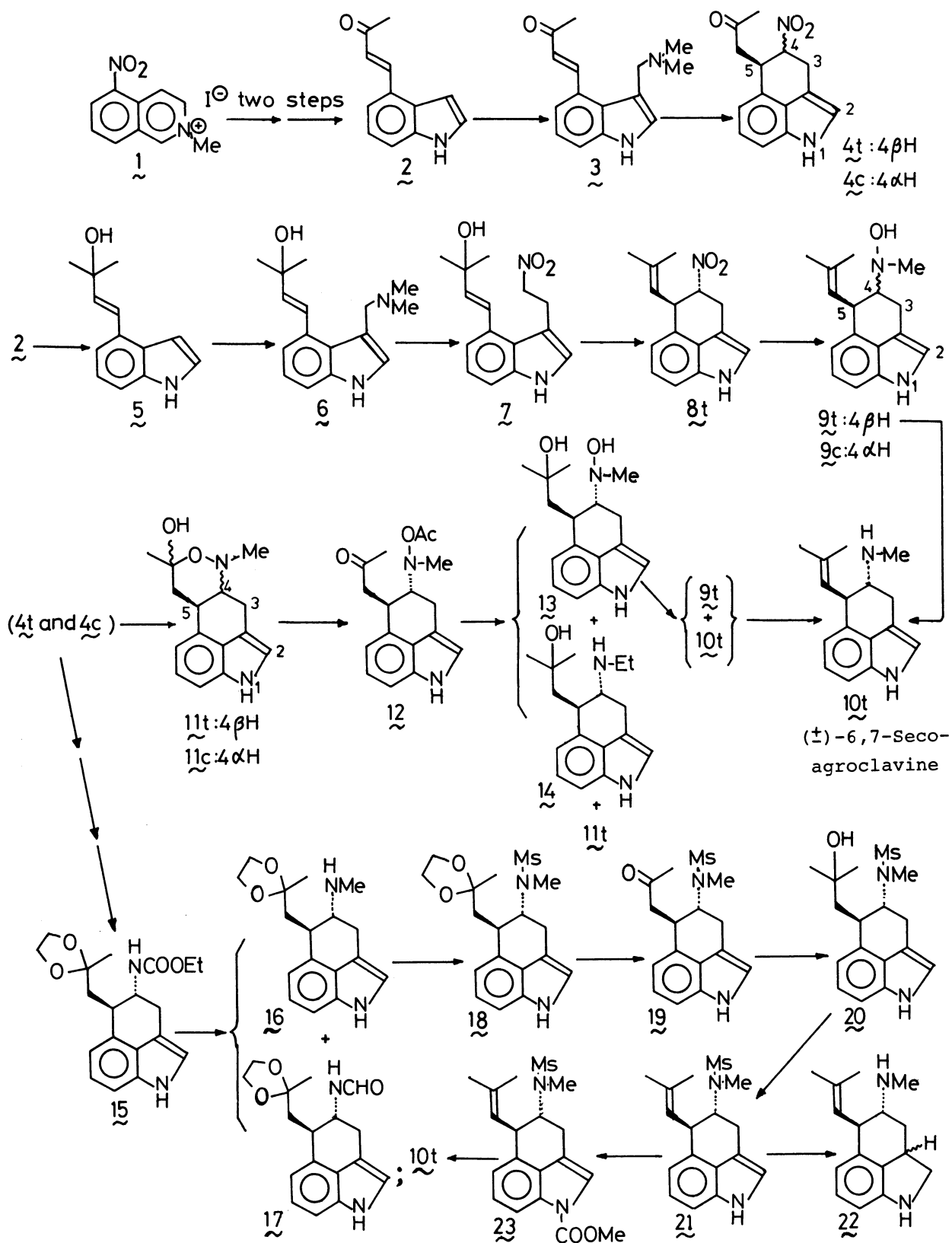
Now, we wish to report a total synthesis of ergot alkaloid, (\pm)-6,7-secoagroclavine³ by three routes starting from 2 or 4t.

I. Eight step synthesis of (\pm)-6,7-secoagroclavine from 1

Grignard reaction of 2 with MeMgI in THF/Et₂O afforded allyl alcohol derivative [5, mp 97-98°C, MS m/e: 201 (M⁺), NMR (CD₃OD) δ : 1.41 (6H, s), 6.37 (1H, d, J=15.6 Hz), 6.90 (1H, d, J=15.6 Hz)] in 84% yield. Subsequent treatment of 5 with dimethyl-(methylene)ammonium chloride⁴ in CH₃CN gave Mannich base [6, oil, MS m/e: 258 (M⁺), NMR (CDCl₃) δ : 2.23 (6H, s), 3.53 (2H, s)] in 70% yield. Monoalkylation of 6 with nitromethane in the presence of (n-Bu)₃P as a catalyst^{2b} afforded nitroallyl alcohol [7, mp 106-107°C, MS m/e: 274 (M⁺), IR (KBr): 1562, 1537, 1382, 1370 cm⁻¹] in 55% yield. The required intra-molecular π -alkylation of allylic moiety of 7 was then achieved by the action of anhydrous ZnCl₂ and NEt₃ in abs. THF to produce 8 in 18% yield, together with 29% yield of the recovery. The NMR spectrum of 8 established predominant formation of trans isomer (trans and cis in ca. 9:1 ratio), whose coupling constant between H-4 and H-5 protons was 9.5 Hz.

Since simple nitroalkanes can be converted to alkylhydroxylamines by the reaction with Grignard reagents,⁵ we have applied this reaction in the next step. Thus, by the action of a large excess of MeMgI in THF/Et₂O to the nitro compound [8t, mp 158-160°C, MS m/e: 256 (M⁺), NMR (CDCl₃) δ : 1.78 (3H, s), 1.83 (3H, s), 5.13 (1H, d, J=10 Hz)], the desired methylhydroxylamines, [9t, oil, MS m/e: 256 (M⁺), NMR (CDCl₃) δ : 2.66 (3H, s, N-Me)] and [9c, oil, MS m/e: 256 (M⁺), NMR (CDCl₃) δ : 2.77 (3H, s, N-Me)], were obtained in 19% and 20% yields, respectively. Final conversion of trans isomer (9t) to the corresponding amine (10t) was achieved in 27% yield by the reduction with aq. TiCl₃⁶ in the presence of AcONH₄.

The synthetic material (10t) was identified as (\pm)-6,7-secoagroclavine by mix-



ture melting point, spectral (IR, UV, MS, and $^1\text{H-NMR}$), and tlc comparison with an authentic sample, which was kindly supplied by Dr. M. Natsume.

II. Nine step synthesis of (\pm)-6,7-secoagroclavine from 1

A mixture of 4t and 4c (in ca. 3:1 ratio) was converted with excess MeMgI in THF/Et₂O to the cyclic hemiketal (11) in 94% yield. The compound (11) was found to be a mixture of 11t and 11c in ca. 2:1 ratio, from which 11t [mp 207-209°C (dec.), MS m/e: 258 (M^+), NMR (10% CD₃OD in CDCl₃) δ : 1.47 (3H, s), 2.74 (3H, s, N-Me)] was separated out by crystallization from MeOH. Subsequent acetylation with Ac₂O/pyridine converted 11t into 12 [mp 136.5-137.5°C, MS m/e: 300 (M^+), IR (KBr): 1733, 1710 cm⁻¹] in 99% yield.

Grignard reaction was again employed on 12 with an excess of MeMgI in THF/Et₂O to produce the desired methylhydroxylamine [13, oil, MS m/e: 274 (M^+), NMR (CDCl₃) δ : 1.36 (6H, s), 2.72 (3H, s, N-Me)], ethylamino compound⁷ [14, mp 185.5-186.5°C, MS m/e: 272 (M^+), NMR (10% CD₃OD in CDCl₃) δ : 1.10 (3H, t, J=7 Hz), 2.73 (2H, q, J=7 Hz)], and 11t in 6%, 16%, and 59% yields, respectively. Although the yield of 13 is low, this difficulty can be overcome by repetition of the above reaction sequences using the recovered 11t. Dehydration of 13 with p-TsOH in benzene afforded a mixture of 9t and 10t.⁸ Since the separation of them was rather difficult, the mixture was immediately reduced with aq. TiCl₃ in the presence of AcONH₄ to produce the alkaloid (10t) in 42% overall yield from 13.

III. Fourteen step synthesis of (\pm)-6,7-secoagroclavine from 1

The ethyl carbamate (15) was prepared from the compound (4t) through three steps as previously reported.^{2b} Reduction of 15 with LiAlH₄ in Et₂O afforded methylamino compound [16, oil, MS m/e: 286 (M^+), NMR (CDCl₃) δ : 2.30 (3H, s, N-Me)]^{3b} and N-formyl compound [17, oil, MS m/e: 300 (M^+), NMR (CDCl₃) δ : 7.78 (1H, s, N-CHO)]^{3b} in 51% and 37% yields, respectively. Treatment of 16 with MsCl in Et₃N/pyridine produced the sulfonamide [18, mp 222-223°C, MS m/e: 364 (M^+), IR (KBr): 1319, 1138 cm⁻¹] in 79% yield. The corresponding ketone [19, mp 166.5-167°C, MS m/e: 320 (M^+), IR (KBr): 1718 cm⁻¹], obtained in 86% yield after deketalization with 1N-HCl in acetone, was then allowed to react with MeMgI to afford the tertiary alcohol [20, mp 194-195°C, MS m/e: 336 (M^+), NMR (CDCl₃) δ : 1.35 (3H, s), 1.47 (3H, s)] in 85% yield. Dehydration of 20 with p-TsOH in benzene gave the olefin [21, mp 171-172°C, MS m/e: 318 (M^+), NMR (CDCl₃) δ : 1.83 (3H, d, J=1.6 Hz), 1.90 (3H, d, J=1.6 Hz), 5.20 (1H, d, J=10 Hz)] in 85% yield.

It is noteworthy that in the final step an attempted demesylation of 21 by Birch reduction resulted in the formation of the indoline [22, in ca. 1:1 mixture of diastereoisomers, MS m/e: 242 (M^+)] regardless of the presence or absence of proton sources under various reaction conditions.⁹ This undesired over-reduction was prevented by the use of 1-methoxycarbonyl compound [23, oil, MS m/e: 376 (M^+), IR (film): 1740 cm⁻¹, NMR (CDCl₃) δ : 3.98 (3H, s, COOMe)], which was obtained from 21 in 86% yield by the reaction with NaH/ClCOOMe in abs. DMF. Removal of both N-mesyl and 1-methoxycarbonyl groups of 23 was successfully carried out with Li in liq. NH₃ at -33°C and subsequent work-up to afford the alkaloid 10t and 22 in 50% and

37% yields, respectively.

Further study for the synthesis of other ergot alkaloids is currently in progress.

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References and Notes

All new compounds gave satisfactory analytical and spectral data. Efforts to attain the optimum reaction conditions have not been made for any of the reaction steps described.

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7. The formation of 14 would be explained by the initial elimination of AcOH from 12 and subsequent attack of MeMgI on the resultant N-methylene intermediate.
8. We believe that homolytic N-O bond fission of hydroxylamine (13) and subsequent abstraction of hydrogen radical would be the reasonable explanation for the formation of 10t.
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